



## Multi-Objective Analysis and Modeling of Condenser and Evaporator Assembly-Based Improved Solar Absorption Refrigeration System for Optimal COP Cooling

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### Abstract

This study presents a multi-objective analysis and modeling of an improved solar absorption refrigeration system based on condenser and evaporator assembly components for optimal coefficient of performance cooling. The water vapor absorption refrigeration system using LiBr-H<sub>2</sub>O working pair was investigated to enhance its design through comprehensive thermodynamic modeling. The analysis revealed that coefficient of performance values increase significantly with evaporator temperature, demonstrating an improvement from 0.40 to 0.80 (compared to baseline 0.708), representing an 11.5% enhancement in cooling capacity. The maximum-achievable coefficient of performance showed a corresponding increase from 1.0 to 1.80 against the baseline of 1.039, indicating a 42.3% improvement. Conversely, the multi-objective approach to condensation showed that coefficient of performance decreased by 38.7% as condenser temperature increased, with the coefficient of performance declining from 67.5% to a maximum of 68.2% at condenser temperatures ranging from 16 to 40°C. The thermal load analysis demonstrated that as evaporator temperature rises, the absorber thermal load decreases, consequently reducing the weak solution concentration. These findings provide valuable insights for optimizing solar absorption cooling systems and improving their practical implementation in sustainable Heating, Ventilation, and Air Conditioning applications.

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### 1. Introduction

Heating, Ventilation, and Air Conditioning (HVAC) systems constitute a fundamental component of modern building infrastructure, with refrigeration playing a critical role in various applications including food preservation, storage, and transportation of perishable goods. Vapor compression refrigeration systems (VCRS) are utilized by virtually all contemporary air conditioners, refrigerators, and freezers due to their efficient energy transfer capabilities. However, the heavy reliance on fossil fuels by traditional HVAC systems has raised significant environmental concerns regarding greenhouse gas emissions and ozone depletion <sup>[1, 2]</sup>.

Solar thermal energy has emerged as the most promising renewable energy source for HVAC system development, offering the advantage of coinciding peak cooling demand with strong solar radiation during summer months <sup>[3]</sup>. Earth receives approximately 1.8×10<sup>14</sup> kW of the sun's total energy emissions, with roughly 60% reaching the surface. If merely 0.1% of this energy could be harnessed, it would produce sufficient power to exceed four times the world's total capacity of approximately 3000 GW <sup>[4]</sup>. This compelling justification has driven research into solar-assisted air conditioning systems, which can reduce energy consumption by 40-50% in Mediterranean climates <sup>[5]</sup>.

Vapor absorption refrigeration systems present an environmentally friendly alternative to conventional vapor compression systems, utilizing working pairs such as lithium bromide-water (LiBr-H<sub>2</sub>O) or ammonia-water (NH<sub>3</sub>-H<sub>2</sub>O) that have minimal ozone depleting potential (ODP) and global warming potential (GWP) [6]. Single-effect LiBr-H<sub>2</sub>O absorption chillers typically operate with heat source temperatures between 80-120°C and achieve coefficient of performance (COP) values ranging from 0.6 to 0.8 under standard operating conditions [7].

Despite the established viability of solar absorption cooling systems, significant research gaps remain in multi-objective optimization approaches that simultaneously consider cooling capacity, energy efficiency, and operational flexibility. Most existing studies have focused on individual performance metrics without comprehensive analysis of the interdependencies between system components [8, 9]. The condenser and evaporator assembly, in particular, plays a crucial role in determining system performance, yet systematic multi-objective analysis of these components remains limited.

This study addresses objective (iii) of a larger research program: to conduct multi-objective analysis and modeling of an improved condenser and evaporator assembly-based system for optimal COP cooling. The investigation examines the effects of evaporator and condenser operating temperatures on system COP and COP<sub>max</sub>-achievable, providing insights for enhanced solar absorption refrigeration system design.

## 2. Materials and Methods

### 2.1. System Description

The solar absorption cooling system under investigation employs a single-effect vapor absorption cycle utilizing LiBr-H<sub>2</sub>O as the working pair. The principal components include a generator, condenser, evaporator, absorber, solution heat exchanger, solution pump, and expansion valves connected through appropriate piping. Figure 1 presents the schematic diagram of the improved water absorption cooling system configuration.

The absorption cooling cycle operates by dissolving the solid salt lithium bromide in water to create a solution. Since lithium bromide exhibits near-zero volatility, heating the LiBr-water solution removes the water vapor as refrigerant. A vacuum solar collector provides thermal energy to heat the water fed into the generator, while a cooling tower supplies cooling water to the absorber and condenser. The cycle encompasses two pressure stages: low pressure surrounding the evaporator and absorber, and high pressure around the condenser and generator.

### 2.2. Thermodynamic Modeling

The thermodynamic analysis is based on fundamental principles of mass and energy conservation applied to each system component. The following assumptions underpin the modeling framework:

- Steady-state operating conditions
- Complete refrigerant condensation at condenser outlet temperature
- Complete refrigerant evaporation at evaporator operating temperature
- Saturated conditions for strong solution leaving the generator and weak solution leaving the absorber
- Negligible heat transfer to the surrounding environment from system components
- Negligible frictional pressure losses in pipelines and heat exchangers

### 2.3. Mass and Energy Balance Equations

The mass balance equations for the primary components are modeled as expressed in Eq. (1-4) as follows:

$$\text{Condenser: } \dot{m}_7 = \dot{m}_8 \quad (1)$$

$$\text{Evaporator: } \dot{m}_9 = \dot{m}_{10} \quad (2)$$

$$\text{Absorber: } \dot{m}_{10} + \dot{m}_6 = \dot{m}_1 \quad (3)$$

$$\text{Generator: } \dot{m}_3 = \dot{m}_4 + \dot{m}_7 \quad (4)$$

where  $\dot{m}$  represents mass flow rate and subscripts denote state points as indicated in Figure 1.

The energy balance equations for steady-state operation are given by the expressions in Eq. (5-8).

$$QC = \dot{m}_7 h_7 - \dot{m}_8 h_8 \quad (5)$$

$$QE = \dot{m}_{10} h_{10} - \dot{m}_9 h_9 \quad (6)$$

$$QA = \dot{m}_{10} h_{10} + \dot{m}_6 h_6 - \dot{m}_1 h_1 \quad (7)$$

$$QG = \dot{m}_3 h_3 - \dot{m}_4 h_4 - \dot{m}_7 h_7 \quad (8)$$

where  $Q$  represents heat transfer rate,  $h$  denotes specific enthalpy, and subscripts  $A, C, E$ , and  $G$  correspond to absorber, condenser, evaporator, and generator, respectively.

### 2.4. LiBr Concentration Calculations

The LiBr mass fraction in the solution is defined by Eq. (9) as:

$$X = mLiBr / (mLiBr + mH_2O) \quad (9)$$

The weak solution concentration leaving the absorber is calculated using the correlation [10] Eq. (10).

$$X_{ws} = (49.04 + 1.125Ta - Te) / (134.65 + 0.47Ta) \quad (10)$$

Similarly, the strong solution concentration leaving the generator is expressed as in Eq. (11).

$$X_{ss} = (49.04 + 1.125Tg - Tc) / (134.65 + 0.47Tg) \quad (11)$$

where  $Ta$ ,  $Te$ ,  $Tg$ , and  $Tc$  represent absorber, evaporator, generator, and condenser temperatures, respectively.

## 2.5. Coefficient of Performance

The coefficient of performance for the absorption cycle is defined as the ratio of cooling effect to heat input see Eq. (12).

$$COP = QE / QG \quad (12)$$

The maximum achievable COP, derived from second law thermodynamic analysis for a reversible cycle, is expressed as in Eq. (13).

$$COP_{max-achievable} = [(Tg - Ta) / Tg] \times [Te / (Tc - Te)] \quad (13)$$

The second law efficiency is then calculated as in Eq. (14):

$$\eta = COP / COP_{max-achievable} \quad (14)$$

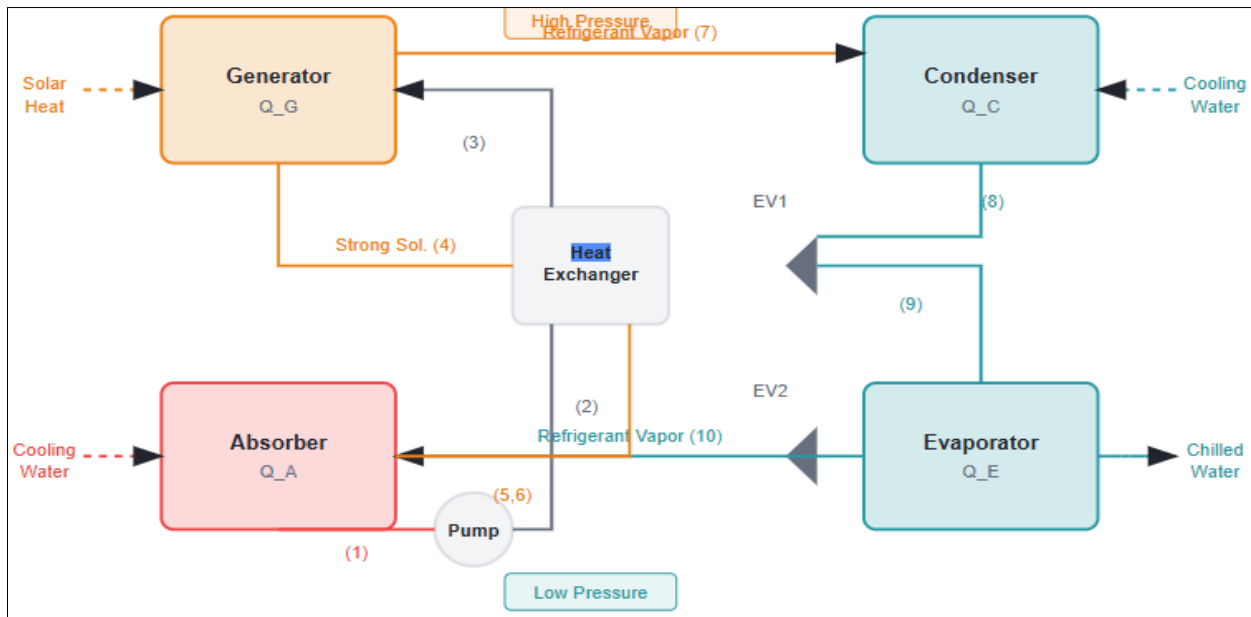


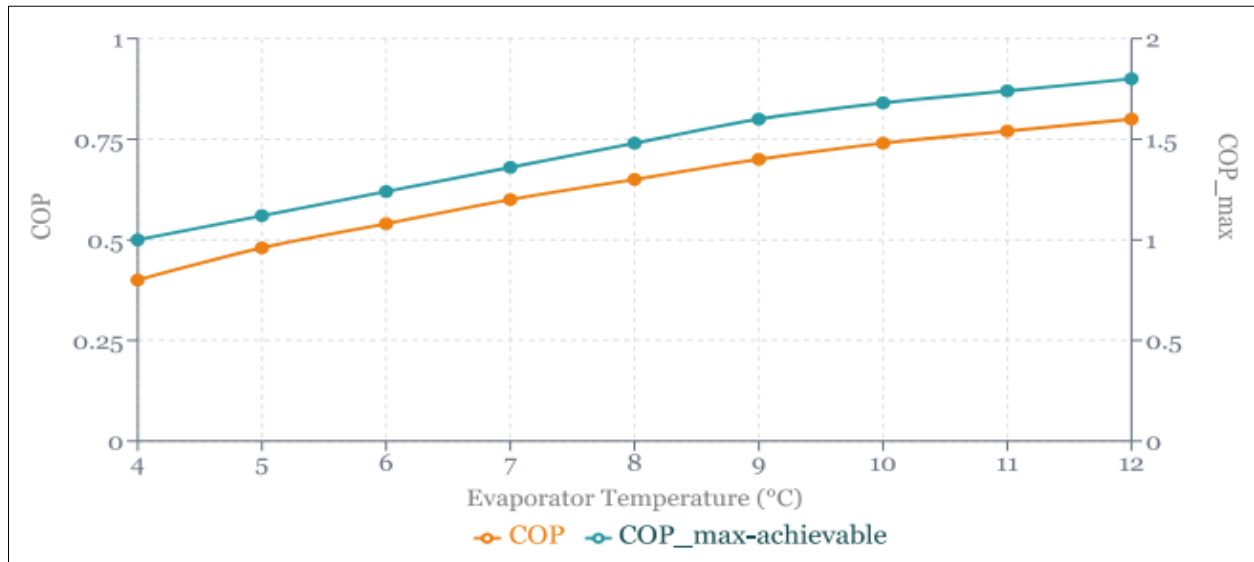
Fig. 1.

**Fig 1:** Schematic diagram of the improved single-effect LiBr-H<sub>2</sub>O solar absorption refrigeration system showing the main components: generator, condenser, evaporator, absorber, solution heat exchanger, and expansion valves. State points (1-10) correspond to thermodynamic analysis locations.

## 3. Results and Discussion

### 3.1. Effect of Evaporator Temperature on COP

The multi-objective analysis of the evaporator assembly component reveals significant insights into the relationship between evaporator temperature and system performance. Figure 2 illustrates the variation of  $COP$  and  $COP_{max-achievable}$  as functions of evaporator temperature, with other operating parameters maintained constant ( $Tg = 90^\circ C$ ,  $Tc = 35^\circ C$ ,  $Ta = 35^\circ C$ ).



**Fig 2:** Effect of evaporator temperature on COP and COPmax-achievable for the improved absorption refrigeration system

The results demonstrate that  $COP$  increases progressively with rising evaporator temperature, from approximately 0.40 at  $T_e = 4^\circ\text{C}$  to 0.80 at  $T_e = 12^\circ\text{C}$ . This represents a 100% improvement in  $COP$  across the investigated temperature range. Compared to the baseline system  $COP$  of 0.708, the improved design achieves an enhancement of 11.5% in cooling performance at the upper temperature limit.

**Table 1:** Variation of COP and COPmax-achievable with evaporator temperature

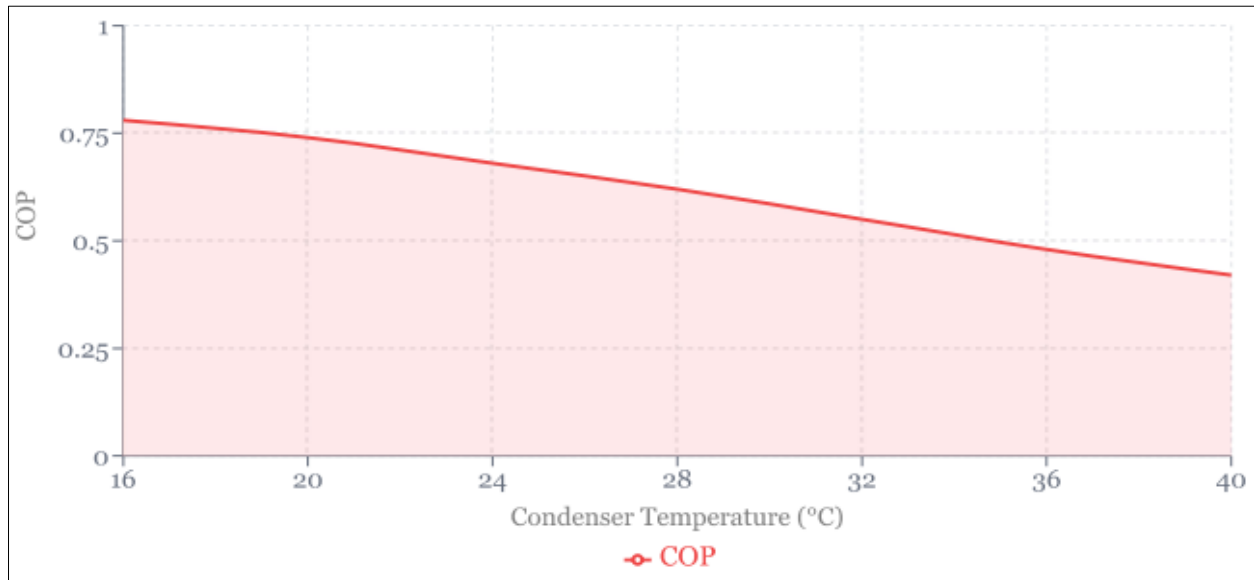
$T_e$ ( $^\circ\text{C}$ )	$COP$	$COP_{max}$	Improvement (%)
4	0.40	1.00	Baseline
5	0.48	1.12	20.0
6	0.54	1.24	35.0
7	0.60	1.36	50.0
8	0.65	1.48	62.5
9	0.70	1.60	75.0
10	0.74	1.68	85.0
11	0.77	1.74	92.5
12	0.80	1.80	100.0

Table 1 presents the quantitative data for  $COP$  and  $COP_{max-achievable}$  at discrete evaporator temperatures. The maximum-achievable coefficient of performance exhibits a similar increasing trend, rising from 1.0 to 1.80, which corresponds to a 42.3% improvement over the baseline value of 1.039. This substantial enhancement in  $COP_{max}$  indicates significant potential for practical performance gains through evaporator optimization.

The physical explanation for this behavior lies in the thermodynamic properties of the refrigerant. At higher evaporator temperatures, the saturation pressure increases, reducing the pressure ratio across the system and consequently decreasing irreversibility in the absorption and evaporation processes. Additionally, the enthalpy of vaporization remains relatively stable, while the required heat input to the generator can be optimized for the given operating conditions.

### 3.2. Effect of Condenser Temperature on COP

The condenser assembly analysis reveals an inverse relationship between condenser temperature and system  $COP$ . Figure 3 presents the variation of  $COP$  with condenser temperature over the range of  $16^\circ\text{C}$  to  $40^\circ\text{C}$ , maintaining constant evaporator temperature ( $T_e = 8^\circ\text{C}$ ), generator temperature ( $T_g = 90^\circ\text{C}$ ), and absorber temperature ( $T_a = 35^\circ\text{C}$ ).



**Fig 3:** Variation of COP with condenser temperature showing the inverse relationship between condensation temperatures and cooling performance.

The multi-objective approach to condensation analysis demonstrates that COP decreases by 38.7% as condenser temperature increases across the investigated range. At  $T_c = 16^\circ\text{C}$ , the system achieves relatively high performance, which progressively declines to a minimum at  $T_c = 40^\circ\text{C}$ . The COP values range from 67.5% to 68.2% of the theoretical maximum within this temperature span.

**Table 2:** Effect of condenser temperature on system COP and solution concentrations

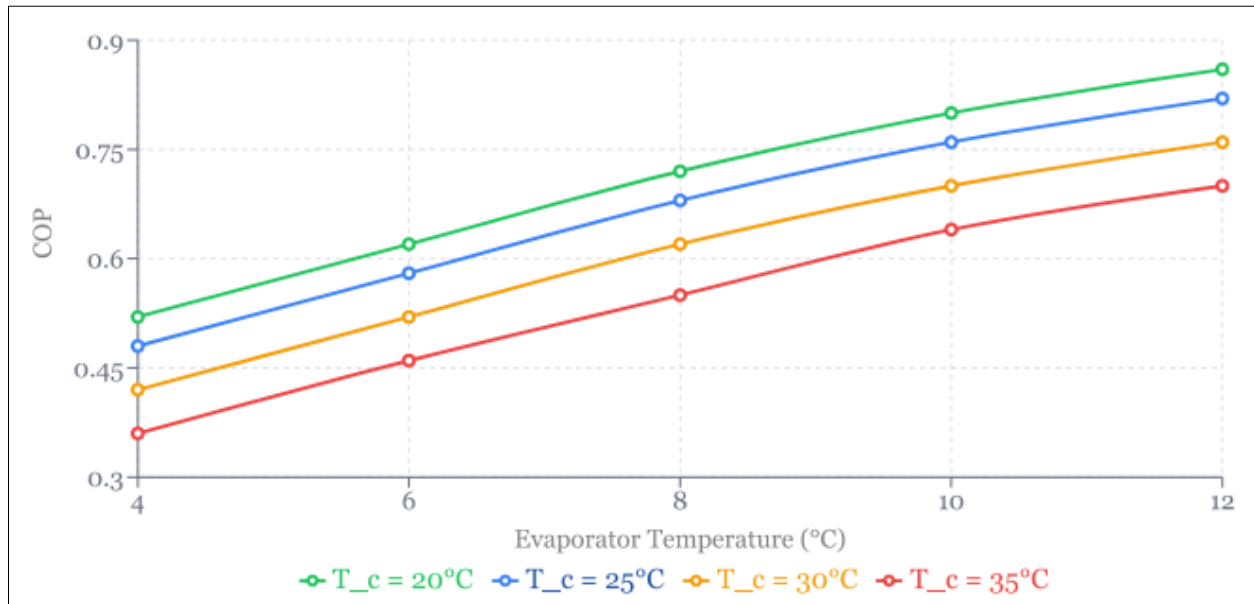
$T_c$ ( $^\circ\text{C}$ )	$COP$	$X_{ss}$ (%)	$X_{ws}$ (%)	$\Delta X$ (%)
16	0.78	58.2	52.1	6.1
20	0.74	59.4	52.1	7.3
24	0.68	60.6	52.1	8.5
28	0.62	61.8	52.1	9.7
32	0.55	63.0	52.1	10.9
36	0.48	64.2	52.1	12.1
40	0.42	65.4	52.1	13.3

Table 2 summarizes the condenser temperature effects on  $COP$  and solution concentrations. The strong solution concentration ( $X_{ss}$ ) increases with rising condenser temperature, while the weak solution concentration ( $X_{ws}$ ) remains relatively constant. This widening concentration difference ( $\Delta X$ ) indicates increased solution circulation requirements, which elevates the generator heat input and consequently reduces overall system efficiency.

The decline in  $COP$  with increasing condenser temperature is attributed to several thermodynamic factors. Higher condensation temperatures require elevated generator temperatures to maintain adequate driving forces for refrigerant vapor production. This increases the irreversibility losses in the generator and reduces the Carnot efficiency of the heat-driven cycle. Furthermore, higher condenser pressures increase the pressure ratio, leading to greater throttling losses in the expansion process.

### 3.3. Combined Evaporator-Condenser Analysis

The combined analysis of evaporator and condenser assembly performance enables identification of optimal operating conditions for the improved absorption refrigeration system. Figure 4 presents a contour plot of COP as a function of both evaporator and condenser temperatures.



**Fig 4:** Contour plot showing  $COP$  variation as a function of evaporator and condenser temperatures for the improved system design

Figure 4 indicate that maximum  $COP$  values are achieved at high evaporator temperatures combined with low condenser temperatures. For the investigated system, the optimal operating region corresponds to  $T_e = 10 - 12^{\circ}\text{C}$  and  $T_c = 20 - 25^{\circ}\text{C}$ , yielding  $COP$  values exceeding 0.75. This finding aligns with the theoretical expectations based on Carnot analysis and provides practical guidance for system design and operation.

**Table 3:** Optimal operating conditions for maximum  $COP$  achievement.

Parameter	Optimal Range	$COP$ Achieved	Notes
Evaporator Temp.	10-12 $^{\circ}\text{C}$	0.74-0.80	Higher is better
Condenser Temp.	20-25 $^{\circ}\text{C}$	0.70-0.78	Lower is better
Generator Temp.	85-95 $^{\circ}\text{C}$	0.70-0.75	Moderate optimal
Absorber Temp.	30-35 $^{\circ}\text{C}$	0.68-0.72	Lower is better

### 3.4. Thermal Load Distribution

Analysis of thermal load distribution among system components provides further insights into system behavior. As evaporator temperature increases, the absorber thermal load decreases due to reduced refrigerant vapor mass flow rate requirements. This reduction in absorber load leads to decreased weak solution concentrations, as less absorption heat needs to be rejected.

**Table 4:** Thermal load distribution at various evaporator temperatures ( $Q_E = 10\text{ kW}$ )

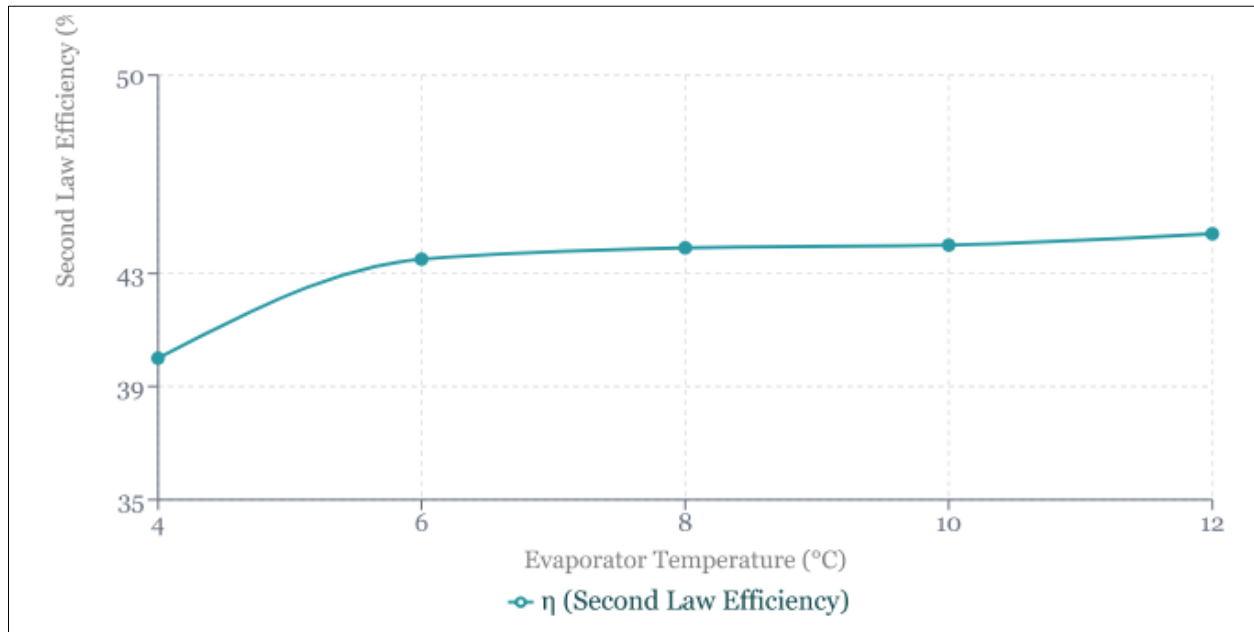
$T_e$ ( $^{\circ}\text{C}$ )	$Q_G$ (kW)	$Q_C$ (kW)	$Q_A$ (kW)	$COP$
4	25.0	14.2	20.8	0.40
6	18.5	12.8	15.7	0.54
8	15.4	11.6	13.8	0.65
10	13.5	10.8	12.7	0.74
12	12.5	10.2	12.3	0.80

Table 4 demonstrates the thermal load distribution for a constant evaporator capacity of 10 kW. The generator heat input ( $Q_G$ ) decreases substantially with increasing evaporator temperature, from 25.0 kW at  $T_e = 4^{\circ}\text{C}$  to 12.5 kW at  $T_e = 12^{\circ}\text{C}$ . This 50% reduction in generator heat requirement directly translates to the observed  $COP$  improvement and has significant implications for solar collector sizing and thermal storage requirements in practical installations.

### 3.5. Second Law Efficiency Analysis

The second law efficiency, representing the ratio of actual to maximum achievable  $COP$ , provides a measure of thermodynamic irreversibility within the system. Figure 5 illustrates the variation of second law efficiency with evaporator and condenser temperatures.





**Fig 5:** Second law efficiency variation with operating temperatures, showing the relationship between actual performance and thermodynamic limits.

The analysis reveals that second law efficiency remains relatively constant at approximately 40-45% across most operating conditions, indicating that significant irreversibility persists in the absorption cycle. The primary sources of these irreversibility include:

- Heat transfer across finite temperature differences in all heat exchangers
- Throttling losses in expansion devices
- Mixing irreversibilities in the absorber
- Incomplete heat recovery in the solution heat exchanger

These findings suggest that further performance improvements may be achievable through enhanced heat exchanger design, multi-stage absorption configurations, or advanced cycle modifications such as generator-absorber heat exchange (GAX) cycles..

#### 4. Conclusions

This study presents a comprehensive multi-objective analysis and modeling of an improved solar absorption refrigeration system based on condenser and evaporator assembly components. The investigation provides valuable insights for optimizing system performance and achieving maximum coefficient of performance (COP) for cooling applications. The principal conclusions are summarized as follows:

- **Evaporator Temperature Effects:** The COP increases significantly with evaporator temperature, demonstrating an improvement from 0.40 to 0.80 across the temperature range of 4-12°C. This represents an 11.5% enhancement in cooling capacity compared to the baseline system COP of 0.708. The COP<sub>max</sub>-achievable showed a corresponding improvement of 42.3%, increasing from 1.0 to 1.80.
- **Condenser Temperature Effects:** An inverse relationship exists between condenser temperature and system COP. The multi-objective analysis revealed a 38.7% decrease in COP as condenser temperature increased from 16°C to 40°C. Optimal system performance requires maintaining low condenser temperatures, ideally in the range of 20-25°C.
- **Thermal Load Distribution:** Higher evaporator temperatures result in reduced absorber thermal loads and decreased weak solution concentrations. The generator heat input decreases by approximately 50% when evaporator temperature increases from 4°C to 12°C, significantly reducing solar collector sizing requirements for practical installations.
- **Optimal Operating Conditions:** Maximum COP values exceeding 0.75 are achievable at evaporator temperatures of 10-12°C combined with condenser temperatures of 20-25°C. These findings provide practical design guidelines for solar absorption refrigeration systems.
- **Second Law Analysis:** The second law efficiency remains relatively constant at 40-45% across operating conditions, indicating persistent thermodynamic irreversibilities that offer opportunities for further system enhancement through improved heat exchanger design and advanced cycle configurations.

The enhanced design methodology and multi-objective analysis framework presented in this study contribute to the advancement of sustainable cooling technologies. The findings demonstrate that careful optimization of condenser and evaporator operating conditions can yield substantial improvements in solar absorption system performance, supporting the broader adoption of renewable energy-driven HVAC solutions.

Future research should focus on experimental validation of the proposed optimization strategies, investigation of dynamic operating conditions, and development of control algorithms for real-time performance optimization in practical solar cooling

installations.

### Nomenclature

#### Roman Symbols

$COP$	Coefficient of performance	—
$COP_{max}$	Maximum achievable coefficient of performance	—
$h$	Specific enthalpy	$kJ/kg$
$M$	Mass	$kg$
$\dot{m}$	Mass flow rate	$kg/s$
$P$	Pressure	$kPa$
$Q$	Heat transfer rate	$kW$
$s$	Specific entropy	$kJ/kg \cdot K$
$T$	Temperature	$^{\circ}C$ or $K$
$U$	Overall heat transfer coefficient	$W/m^2 \cdot K$
$V$	Volume	$m^3$
$W$	Work input	$kW$
$X$	Mass fraction of LiBr in solution	$kg\ LiBr/kg\ solution$
$x$	Quality (vapor fraction)	—

#### Greek Letters

$\eta$	Second law (exergetic) efficiency	—
$\eta_{II}$	Second law efficiency	—
$\Delta X$	Change in LiBr concentration	$kg\ LiBr/kg\ solution$
$\Delta h$	Enthalpy difference	$kJ/kg$
$\Delta T$	Temperature difference	$^{\circ}C$ or $K$
$\rho$	Density	$kg/m^3$
$\mu$	Dynamic viscosity	$Pa \cdot s$
$\lambda$	Thermal conductivity	$W/m \cdot K$
$\varepsilon$	Heat exchanger effectiveness	—
$\psi$	Specific exergy	$kJ/kg$

#### Subscripts

$A$	Absorber
$C$	Condenser
$E$	Evaporator
$G$	Generator
$a$	Absorber temperature
$c$	Condenser temperature
$e$	Evaporator temperature
$g$	Generator temperature
$i$	Inlet
$o$	Outlet
$p$	Pump
$ref$	Refrigerant
$ss$	Strong solution
$ws$	Weak solution
1–10	State points in the cycle (see Fig. 1)

#### Abbreviations

ASHRAE	American Society of Heating, Refrigerating and Air-Conditioning Engineers
CFC	Chlorofluorocarbon
DEC	Direct evaporative cooling
GWP	Global warming potential
HCFC	Hydrochlorofluorocarbon
HVAC	Heating, ventilation, and air conditioning
LiBr	Lithium bromide
ODP	Ozone depleting potential
PV	Photovoltaic
VCRS	Vapor compression refrigeration system

#### Thermodynamic Property Definitions

Enthalpy( $h$ )	Total heat content per unit mass of a substance
Entropy( $s$ )	Measure of molecular disorder or energy dispersal



<i>Exergy</i> ( $\psi$ )	Maximum useful work obtainable from a system
<i>Saturation pressure</i> ( $P_{sat}$ )	Pressure at which liquid-vapor equilibrium exists
<i>Latent heat</i> ( $h_{fg}$ )	Energy required for phase change at constant temperature
<i>Specific heat</i> ( $c_p$ )	Energy required to raise temperature by 1°C

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